



Practical chemistry, with Notes and questions on theoretical chemistry

By William Ripper

RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 36 pages. Dimensions: 9.7in. x 7.4in. x 0.1in. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1883 Excerpt: . . . 11-16 litres of chlorine can be obtained from 86-5 grams of hydrochloric acid, how many litres of chlorine can be obtained from 50 grams of that acid grams grams litres 86-5: 50: : 11-16: x 50 x 11-16 15-287 litres of chlorine. To find the volume resulting from the combination of gaseous bodies. Example 1. --If a mixture of 20 c. c. of hydrogen and 10 c. c. of oxygen at 140C. be exploded, what volume of steam will be formed Answer. 20 c. c of steam. Example 2. --If 100 c. c. of hydrogen are mixed with 150 c. c of chlorine, and exploded, what volume of hydrochloric acid is produced Which gas, and how much of it, remains uncombined By the equation--H Cl HC1 hydrogen chlorine hydrochloric acid. and 1 vol. 1 vol. 2 vols, therefore 100 c. c. 100 c. c. 200 c....



READ ONLINE

[2.96 MB]

Reviews

The book is straightforward in read safer to recognize. This really is for anyone who statte there had not been a worthy of looking at. You may like just how the blogger create this publication.

-- **Friedrich Nolan**

It in just one of the best publication. This can be for anyone who statte that there was not a well worth reading through. Once you begin to read the book, it is extremely difficult to leave it before concluding.

-- **Tara Jerde**